

2013

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3.1.2.		μ		81
3.1.3			μ	83
3.1.4	μ		μ	89
3.2				90
3.2.1			μ	90
3.2.2	μ			92
				95
4.1	μ		μ	95
4.1.1				96
4.1.2				96
4.1.3	μ			97
4.1.4				98
4.1.5				100
4.1.6				101
4.1.7				103
4.2			μ	104
				104
4.2.1	-		μ	105
4.2.2				107
4.2.3				108
4.3		μ		110
4.3.1		μ	μ	110

4.3.2		111
4.4			.112
4.4.1		112
4.4.2	Mg	113
4.4.3	Ca	114
4.4.4		115
4.4.5	Mg	116
4.4.6	Ca	117
-		119
		130

μ μ . μ
μ 2-10 mm μ μ .
μ μ μ μ . μ
μ μ μ , μ μ μ
μ μ μ . μ μ
μ μ μ , μ μ
μ μ μ μ μ μ
μ μ μ (2-10) μ
μ μ μ μ μ
μ μ μ μ μ
μ μ μ μ μ μ
μ μ μ μ μ μ
μ μ μ μ μ μ

ABSTRACT

The development of modern greenhouse complexes in our country and the capability to grow more than one cultivation per year, created new data, thus approaching the Dutch model even more. The most popular plant cultivation on substrates, is the off-field cultivation on bagged substrates. This post-graduate research examined the growth of tomato plants cultivated on both local and imported substrates, which are commonly used in the country during the last decades.

The subject of this research was the evaluation of each substrate's effect on plant growth and nutritious condition and on fruit quality characteristics. The five most commonly used substrates were examined: Rockwool, perlite, coir, pumice (2-10) and pumice (0-8). The cultivated tomato variety was the 'Primadonna' hybrid, which is considered to be very productive, grafted onto 'Maxi-Fort' rootstock.

The experiment took place in the glass greenhouse of the Lab. Different 'recipes' of nutrient water-based solutions were tested on the plants, both during the growing and the reproductive stages of the plant. The irrigation frequency and quantity were the same for all the tested substrates and were adapted to the plant's needs and to the environmental conditions.

During the initial stages of the cultivation, a random plant selection took place, in order to compare the initial effects of the different substrates on the growing plants. Leaf and shoot samples were collected to be tested for their fresh and dry weight and their dry matter content percentage. The plants cultivated on rockwool and coir showed the best growth and those cultivated on perlite the least. The leaf width and plant height measurements improved the results about the development of plant growth on each substrate. During the end of the cultivation, leaf and shoot samples were collected to be tested about their nutritious condition. Plants grown on rockwool (2-10) showed a significantly higher magnesium and calcium content, compared to the plants grown on the rest substrates.

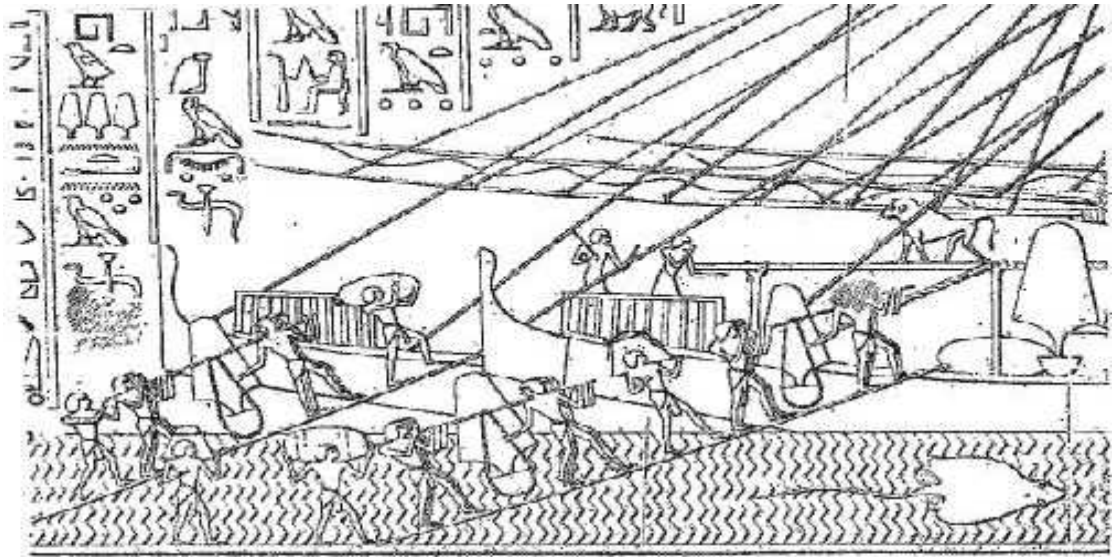
The total fruit weight collected and the number of fruits per substrate were measured. The results did not show significant differences between the examined substrates, but fruit samples collected from plants grown on coir showed increased

weight. Moreover, during the tests that took place in the Lab, the quality characteristics of the fruits collected were examined. The tests showed significant differences considering the total soluble solids content and fruit grown on perlite and coir showed by far increased concentrations compared to fruit grown on the rest substrates. Fruit grown on pumice (2-10) and on perlite were found to be highly firmness.

1. (Soilles Culture)

1.1

μ , μ μ μ μ μ .
μ μ 4000 , μ μ μ
μ μ μ μ μ
(1) (Raviv and Lieth, 2008). μ
μ
(14 -16 μ. .),
μ .
μ μ μ μ μ μ μ (Benton, 2005).
17 μ μ μ μ ,
μ μ μ , μ
μ .
(Raviv and Lieth, 2008).
1946
μ μμ μ μ .
μ
(Cooper, 1975? Verwer, 1976). μ
μ μ μ ,
μ μ 1950 1960
μ , μ μ
(Matkin and Chandler, 1957).



1:

μ

μ

(Navelle, 1913; Matkin et al., 1957).

μ

μ

μ

μ

μ

μ

μ

μ

μ

(Raviv and

Lieth, 2008).

μ

μ

μ

μ

μ

μ

μ

μ

μ

(Raviv

and Lieth, 2008).

μ

Sachs (1859, 1861)

Knop (1860)

μ

μ

, μ

μ

μ

μ

Sachs Knop
 (Eastwood, 1947).
 70,
 Cooper (1975, 1979),
 (Nutrient Film Technique)
 (Verwer, 1976 1978)

1.2 (hydroponics)

hydroponics
 hydro μ ponos
 1930,
 (1929, 1937, 1940)
 1920
 1940 μ
 (Benton, 2005).

μ , μ . μ 20

μ μ μ . 1977 Harris
 μ , μ
 μ , μ , μ μ μ
 , μ μ μ
 , μ μ μ
 μ .

Encyclopedia Americana International Edition (2000)

μ μ μ .
 μ . Resh (1995)
 μ , μ μ , μ ,
 μ μ μ .

Wignarjah (1995)

μ μ
 μ . μ μ
 μ , μ ,
 μ , μ μ μ μ
 μ (Benton, 2005).
 μ μ μ
 μ , μ μ
 μ μ .
 μ
 μ
 (Benton, 2005).

1.3

1980 μ
 μ μ μ . 60.000
 μ μ , μ , μ μ μ

1995, 2004 55.000 μμ μ (Benton, 2005).

μμ . μ μ μ 600.000
 μ μ (2), μ μ
 (), μ μ
 μ μ (NFT). μ μ
 μ μ , μμ (μμ)
 (. .). μ ,
 μ μ μ ()
 (2006).

, μ μ μ ,
 μ μ μ ,
 μ μ μ
 ,
 μ .
 μ μ μ , μ
 μ . μ
 μ μ
 , μ
 μ (, 2003).

2:
 μ
 μ μ
 μ ().
 (: <http://www.hungrycitybook.co.uk>)



, μ
 μ , μ μ
 μ ,
 . ,
 μ μ
 μ .
 μ μ μ μ μ μ (, 2003).
 μ μ
 1.500 . μ μ μ μ ,
 μ 450 . μ
 μ 50 . μ ,
 μ . . μ 150 . μ
 μ μ 75%, 10%, ΟΤΤ
 10% 5%, . . . (Olympios, 2008).

1.4 μ

μ μ
 μ μ .
 μ ,
 μ μ 19
 μ , μ .
 μ μ μ .
 μ , μ μ μ
 μ μ μ μ μ

μ , μ
 μ μ
 .
 μ . 20 μ
 μ μ μ
 μ μ (Sonneveld and Voogt, 2009).

1.5 μ

: μ μ
 μ μ μ
 .
 μ , μ μ
 , μ
 μ . μ
 μ . μ
 μ μ , μ
 , μ
 μ , μ μ μ μ .
 μ μ , μ μ
 μ μ μ μ ,
 μ .
 μ μ μ μ μ
 μ (, , .), μ
 (, μ
 , . . .). , μ μ
 μ μ

(. . . μ NFT, plant plane hydroponics, . . .)

μ μ μ , μ μ μ , μ μ μ , μ μ μ , μ μ μ , μ μ μ .

μ _____

- I. μ
 - i. μ μ μ μ
 - ii. μ NFT
 - iii.
 - iv. Plant plane hydroponics ()
- II. μ
 - i. μμ (sand culture)
 - ii. (gravel culture)
 - iii. μ
 - iv. μ
 - v.
 - vi. μ
 - vii. μ
 - viii. μ
- III. μ μ -



3: μ μ μ μ ()
 μ . μ
 μ .

(: http://www.freshplaza.com/news_detail.asp?id=91613)



4: μ μ

(Floating system).

μ μ μ .

(: <http://float-garden.com/blog/?cat=13>)

1.6

$\mu \quad \mu \quad \mu$

1.6.1

μ

- μ
 $\mu \mu$.
 $\mu \mu$
- μ
 μ .
 μ
- μ
 $\mu \mu$.
 μ .
 μ
- μ
 μ .
 $\mu \mu$.
 μ .
 μ .
 $\mu \mu$
- μ
 μ . FT
 μ .
 μ
- μ
 $\mu \mu$.
 μ .
 μ .
 μ .
 $\mu \mu$

- μ ().
- μ . μ μ .
- μ μ . μ , $\mu\mu$, μ μ .
- μ μ . μ μ .
- μ μ . μ μ .
- μ μ . μ μ .
- μ μ . μ μ .
- μ μ , μ μ .
- pH μ μ .

1.6.2 μ

- $\mu\mu$, μ μ , μ μ .
- μ μ μ .
- μ , μ .
- (pythium, Fusarium) μ μ .

μ pH μ
 μ (2003).
 μ μ μ
 . μ μ μ μ
 , μ
 μ μ μ (Raviv and Lieth, 2008).

1.8.2 μ μ

μ μ μ μ ,
 μ μ μ μ .
 μ (, , μ) μ
 , μ μ μ
 μ μ μ μ μ
 , μ
 μ μ μ μ μ
 μ μ μ μ μ
 μ (, 1998).
 ,

μ μ μ μ μ μ
 . μ μ
 μ μ μ μ μ μ
 . μ μ μ μ μ
 μ μ μ μ μ μ

μ () μ (μ ,
 μμ). μ . μ ,
 μ μ ,
 μ , pH,
 μ (EC) (Raviv and Lieth, 2008).

μ

(1) , μ
 μ , 2-6 % (Alkan
 and Dogan, 1998; Dogan and Alkan, 2004).

μ 1200-1300 C, μ μ
 μ .
 μ μ μ
 μ , μ .
 μ , μ μ
 μ () ,
 μ .
 μ μ .
 μ μ , , , ,
 . .. μ μ .
 μ μ μ

μ (Bures *et al.*, 1997). μ
 μ pH 7,0-7,5, μ μ
 . μ p
 μ (Raviv and Lieth, 2008).

μ 3/4
 (SiO₂), 1/4 (Al₂O₃)
 14%, , , . . .

μ μ μ ,
μ μ μ . μ μ μ
μ μ μ .



5: μ
(μ 0,5-2,5 mm). μ

μ
μ μ , μ μ
I/S H.J. Henriksen and V. Kähler 1909.
μ μ μ
, 1969 (Smith, 1987).
1980 Grodan, μ

μ , μ μ μ
 . μ μ
 : 60% , 20% 20% .
 μ μ , μ μ
 μ 1600 C. μ μ
 μ μ μ μ , μ μ
 μ μ μ μ .
 μ μ μ (,)
 μ μ . μ
 μ . μ μ μ μ
 μ , , μ , , μ
 . μ μ
 μ
 μ

(Papadopoulos *et al.*, 1999).

, μ μ
 μ μ , 0,05-0,1 cm⁻³ μ
 μ 92 98% .
 μ , μ
 . μ
 μ , 4% 1cm , μ
 . μ 10 cm (Smith, 1987).

Van der Gaag and Wever (2005)

. μ μ μ ,
 μ μ μ μ μ
 . μ μ
 μ μ μ
 (Wever van Leeuwen, 1995).
 μ μ , μ
 pH. pH 7,0-8,0
 μ 6,0-6,5. μ
 μ .

μ μ . μ Fe
Fe, Mn, Cu Zn μ Rupp and Dudley (1989)
 μ .
 μ μ μ μ , μ
 μ μ μ . μ
 μ μ , μ μ
 μ μ μ , μ μ
 μ μ , μ μ .
 μ (μ),
 μ μ μ
 μ , μ



6: μ μ 'Grodan' μ μ



7: μ

_____ μ _____ μ

μ

'rhyolitic' μ μ

(Challinor, 1996).

μ μ μ

μ

, , , , , μ , μ

μ 0-3 mm, 0-8 mm 2-10 mm (Raviv and Lieth, 2008).



8: μ 0-8 mm



9: μ 2-10 mm

. μ μ ,
 μ μ , μ μ .
 μ μ $0,4-0,8 \text{ gr cm}^{-3}$ $70-85\%$,
 μ .
 μ , μ μ .
 μ , μ , μ , μ ,
 μ , μ μ (Raviv *et al.*, 1999).
 μ μ 6μ ,
 $0 \text{ } 10 \text{ kPa}$ (Raviv *et al.*, 1999). μ
 μ
 μ .
 μ μ , μ .
 μ pH $2,5$ (Raviv and Lieth, 2008)..

μ
, μ . μ μ
(Challinor, 1996). ,
(Raviv and Lieth, 2008).

1.7.3.2 μ

μ μ μ
μ μ .
μ μ . μ μ
, , (), μ , , ,
, μ μ .
μ μ -
μ μ μ
μ CO₂ (Raviv and Lieth, 2008).
μ
(Calkins *et al.*, 1997).

_____ (coir)

μ μ μ 75%
μ 25% , μ ‘ ’.
μ μ .
μ μ . μ
μ μ ,
1980 μ μ
μ . μ , μ,
(Raviv and Lieth, 2008).

, μ , μ . μ , μ μ .
 , μ , μ μ .
 Na μ μ μ μ μ μ .
 (μ) . μ
 μ . μ ' (de Kreij van
 Leeuwen, 2001). μ
 μ μ μ μ . μ
 μ μ μ . 3-4 μ
 μ μ μ μ . μ
 μ μ μ μ ,
 μ μ μ μ
 (Raviv and Lieth, 2008).



10: μ μ (fiber dust coir).



11: μ

μ (Coir Fibre).

(: <http://www.Indiamart.com>)

μ

μ 4,8-6,9, μ pH, μ μ μ

μ EC

μ ,

μ , μ μ

(μ a, K Cl). μ μ μ , μ

EC Na, Cl μ .

μ μ μ K Na,

μ .

μ μ μ

(Prasad, 1997). μ .

μ , μ

5 mg/L. , μ

μ . μ

Ca μ (CaNO₃).

μ , μ

μ (Raviv

and Lieth, 2008).

, μ μ
 μ μ μ .
 μ μ μ
 μ 1mm μ , μ
 μ (Noguera *et al.*, 2003 , Prasad
 Chualáin, 2004). μ μ μ
 μ ,
 μ μ (Prasad and Ni
 Chualáin, 2004). Fornes *et al.* (2003)
 μ μ , μ μ
 μ .
 μ μ μ
 μ ,
 . Nelson *et al.* .
 μ
 μ μ μ μ ,
 μ μ . Raviv *et*
al. (2001) 19%
 μ μ . μ
 μ μ μ
 μ , μ (30-35 C) (Islam *et al.*, 2002).
 μ
 Garcia Deverde (1994). μ μ
Calendula officinalis, *Solenostemon scutellarioides*, *Schefflera spp*, *Kalanchoe*
spp, μ , μ *Primula spp*.
 μ . (*Fragaria* \times *ananassa*
Duchense) μ μ
 (Lopez-Medina *et al.*, 2004).

1.8

μ μ

μ μ

(μ). μ

μ μ μ μ μ

μ μ μ

(Benton, 2005)..

μ μ , μ

μ μ Ca Mg.

μ μ

Ca Mg, 100 30 mg/L (ppm).

μ μ Na, Cl μ

μ μ

(Benton, 2005)..

μ Verwer Wellman

(1980) Farnhand *et al.* (1985)

μ μ μ μ 1

ΣΤΟΙΧΕΙΟ	ΣΥΓΚΕΝΤΡΩΣΗ , mg/L (ppm)
ΒΟΡΙΟ (B)	<1
ΑΨΗΘΙΟ (Ca)	<200
ΑΝΘΡΑΚΙΚΑ (CO ₃)	<60
ΧΛΩΡΙΟ (Cl)	<70
ΜΑΓΝΗΣΙΟ (Mg)	<60
ΝΑΤΡΙΟ (Na)	<180
ΨΕΥΔΑΡΓΥΡΟΣ (Zn)	<1

1:

μ

(Smith, 1999).

μ μ μ
 μ μ μ ,
 μ . μ μ
 μ μ μ μ μ
 . μ μ μ μ
 μ . μ μμ μ
 (Benton, 2005). μ μ μ
 μ μ (Anon., 1997).
 , μ μ μ
 μ μ
 , μ μ
 .
 Το pH μ μ μ
 μ μ μ
 . μ pH μ
 μ , pH CO₂.
 ,
 μ . μ μ CaS₄
 μ μ μ NaCl
 pH μ μ

1.9.1

μ

μ , pH
 μ (E.C.). p μ
 (H⁺) μ μ
 μ
 , μ μ
 μ μ μ μ
 μ μ
 μ μ
 μ μ μ μ Faulkner
 (1998) :

- μ μ μ
- μ pH
- μ
- μ μ
- μ 1500 ppm 4000 ppm.
- μ (2)
- μ μ 18-24 C
-

1.9.1.1

μ

μ

μ (Electrical Conductivity - E.C.) μ
 ,
 μ μ . μ
 μ μ μ μ

μ . μ μ μ
μ μ , μ μ μ
μ . μ μ μ μ μ
μ . μ μ μ μ μ
μ μ μ μ μ μ
, μ μ μ μ μ
. μ μ μ μ μ
μ , μ μ μ μ μ
μ μ μ μ μ μ

1.9.1.2 pH

pH μ μ μ μ
, , μ μ μ
μ μ μ μ μ μ pH
μ , μ μ μ μ μ
μ pH μ μ μ μ
. μ μ μ μ μ μ
μ μ μ μ μ μ

1.9.1.3 μ μ

μ μ μ , μ μ
 μ . μ μ μ
 μ , μ μ . μ
 μ , μ μ
 , μ μ μ .
 μ μ
(Benton, 2005). μ μ
 μ μ μ
 μ . Tindall *et al.* (1990)
 μ , μ 21 C μ 26,7 C
 μ μ , μ μ
 μ .

1.9.2 μ

μ μ μ
 μ μ μ ,
 μ μ . ,
 μ , μ
100-200 μ μ
 . μ (μ μ
 μ) μ μ μ
 , μ μ
 . μ
 μ , μ μ μ
 μ μ .
 μ Ca(μ μ)₂ CaS μ , μ
 . μ

). μ , μ μ
 μ μ μ μ
 μ , , μ μ μ (1/500)
 μ) μ
 μ (, 2005).
 , μ μ Benton (2005) μ
 μ μ :
 • (,)
 •)
 • μ μ .
 • .
 • pH .
 • .
 • μ
 μ μ μ μ / μ μ
 μ μ μ μ μ
 μ (http://www.ekk.aua.gr/excel/index.htm).
 μ μ μ μ μ μ μ μ
 μ μ μ μ / Savvas and Adamidis
 (1999) Savvas (2001).

Treatment 1: Ca = 4 meq/L											
: -			μ							E.C. 1,71 dS/m	
: μ			Et 1,70 dS/m							E.C. 0,58 dS/m	
μ μ Ca-salinitymodel			pH opt. 5,58							Calc. (kg) 0,217	
μ μ : 07 2007			X: (K) 0,548							1 2,187	
μ : μ μ			Y: (Ca) 0,296							2 μμ 0,228	
			Z: (Mg) 0,150							3 Fe (6%) 0,140	
			R (N/K) 1,598							4 0,952	
			N: (NH ₄ /tot. N) 0,107							5 5,247	
μ V, m ³ A			P: (P ratio) 0,090							6 μ 3,442	
0,1 100			[Ca] _w 4,000 meq/L							7 μ 0,000	
0,1 100			[Fe] _t 15,00 μmol/l							8 μ 6,792	
0,1 100			[Mn] _t 10,00 μmol/l							9 68% 7,816	
0,1 100			[Zn] _t 3,50 μmol/l							10 μ 59,5	
0,1 100			[Cu] _t 0,50 μmol/l							11 μ 38,5	
0,1 400			[B] _t 20,00 μmol/l							12 5,0	
Calculation of C _b			[Mo] _t 0,50 μmol/l							13 Solubor 35,5	
[H ₃ O ⁺] _w 2,34423E-08			μ							14 3,5	
B _w 22,42532971			. HNO ₃ % 68								
[CO ₃ ²⁻]+[HCO ₃ ⁻]+[H ₂ CO ₃] 0,003776081			. H ₃ PO ₄ % 85								
[H ₃ O ⁺] _(n.s.) 2,63027E-06			% Fe -							50	
B _(n.s.) 1,190549703			6							50	
CALCULATIONS									HCO ₃ ⁻ 0,03		
/	C.C.S	C.C.W.	C.A.F.	SO ₄ ²⁻	NO ₃ ⁻	H ₂ PO ₄ ⁻	HCO ₃ ⁻	Cl ⁻	SO ₄ ²⁻ 0,03		
C.A.S.	15,30			2,25	10,45	1,25	0,60	0,80	N. Solution		
C.A.W.		5,95		1,55	0,00	0,00	3,60	0,80	cats	15,35	
A.A.F.			12,39	0,70	10,45	1,25	0,00	0,00	ans	15,35	
Ca ²⁺	4,00	3,80	0,20	0,00	0,20	0,00	0,00	0,00	Water		
Mg ²⁺	2,00	1,30	0,70	0,70	0,00	0,00	0,00	0,00	catw	5,95	
K ⁺	7,30	0,05	7,25	0,00	6,00	1,25	0,00	0,00	anw	5,95	
NH ₄ ⁺	1,25	0,00	1,25	0,00	1,25	0,00	0,00	0,00	Fertilizers		
Na ⁺	0,80	0,80	0,00	0,00	0,00	0,00	0,00	0,00	catf	12,39	
H ⁺	0,00	0,00	3,00	0,00	3,00	0,00	0,00	0,00	anf	12,39	
									W.C. (L) 220		
									(ml) 217		
									(ml) 9483		
									(ml) 2200		
									(ml) 14022		
									(ml) 1		
									(ml) 550		

2: μ / μ μ

1.10

μ

μ

16 μ

9

μ

.

μ

μ

7

μ

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μ

μ

(, 2012).

1.10.1

1.10.1.1 (N)

μ

,

μ

μ

.

1.10.1.2 (P)

μ 0,5 1%
 μ
 μ
 μ μ μ , μ
 μ μ μ μ μ
 μ
 μ μ μ , μ μ
 μ
 μ μ

(Sonneveld and Voogt, 2009).

μ
 μ
 μ μ
 μ μ
 μ μ μ 1%
 1998b). $\mu\mu$ (Jones
 Mn Zn. μ μ Fe,
 μ μ μ μ μ
 μ μ
 μ 30 50 mg/L , μ 10-20
 mg/L. μ μ μ μ pH
 μ μ μ μ μ μ , μ
 μ
 μ (Benton, 2005).

μ μ $\mu\mu$, μ
 μ μ μ μ μ
 μ μ μ μ
 $\mu\mu$ μ $\mu\mu$
 μ , μ
 pH

, C₃ μ . μ
 μ μ
 μ μ ,
 μ μ
 μ μ
 Mg μ μ Mn μ
 μ μ μ ,
 μ , .
 μ 50
 mg/L, μ μ
 μ μ
 μ μ
 (Benton, 2005).

1.10.1.6

0,15-1% .
 μ , μ .
 Cruciferae Leguminosae μ .
 S μ .
 μ μ ,
 μ μ μ μ .
 μ μ /S
 (Benton, 2005).
 μ μ 50 mg/L,
 μ μ
 μ (, 2012)

1.10.2

μ ,

μ μ 1/10.000

μ . μ

μ μ μ . μ

μ μ μ , μ

μ μ μ μ .

μ μ . Fe,

Mn Zn μ μ pH

μ μ .

1.10.2.1 ()

10-50 mg/kg

μ

, μμ .

μ , μ

μ μ μ .

, μ μ

μ μ μ . μ

μ μ μ .

μ μ μ μ μ

μ μ μ . ,

μ μ μ .

μ μ μ .

, μ μ μ

1.10.2.2 (Fe)

50 100

mg/kg

μ 'phytoferritin'. μ
, μ 'Fe'
(Jones, 2001). Fe μ
 μ ,
 $\mu\mu$ μ μ
(Benton, 2005).
 μ μ μ μ
 μ . μ
, μ μ
Mg. μ μ μ
S, Mn Zn.
Fe
 μ μ μ μ
 . μ μ μ
, μ (Sonneveld and
Voogt, 2009).
 μ
 μ μ .
Fe μ
 μ μ μ (, 2012).

1.10.2.3 (Mn)

20-100 mg/kg

. μ μ ,
 μ μ
, μ μ .

μ μ μ μ μ μ
 μ μ μ μ
 Mn .
 μ 0,5 mg/L (Benton,
 2005).
 μ μ μ n
 μ (, 2012).

1.10.2.4 (Zn)

μ 20 50 mg/kg μ μ
 μ μ μ
 μ μ μ μ μ μ
 , μ μ μ μ μ μ
 . μ
 μ
 Zn
 μ
 μ Zn (, 2012).

1.10.2.5 (Mo)

μ , μ μ μ μ μ μ
 μ μ μ μ μ μ
 Mo μ μ μ μ μ μ
 ,
Cruciferae μ μ .

1.11

μ

μ μ

,

μ

μ

μ ,

μ

μ

μ

,

μ

μ μ

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μ μ

μ

μ

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μ

μ

μ

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μ

μ

μ

μ

μ

μ

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μ

μ .

μ

μ

μ

μ

μ μ

μ

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μ

μ

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μ

μ

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μ

μ

μ

μ

μ

μ

μ

(Raviv and Lieth, 2008).

Kramer *et al.* (2002)

μ

μμ μ

μ

μ

μ

0,2 mm/h.

μ

μ

μ

μ

.

,

μ

μ

μ

(

,

2012).

μ

μ

,

μ

μ

.

μ

Van Assche Vangheed

(1994)

μ

,

μ

.

μ

μ

(100-110 °C 10 min)

μ

.

μ () μ
 μ . μ μ ,
 μ μ , μ μ μ ,
 μ , , .
 , μ , μ (, , μ)
 Jenkins (1948), μ μ
 μ Vera Cruz-Puebla.
 μ (Solanaceae),
 μ ()
 . μ μ
 , μ
 (μ)
 . μ () μ
 , μ .
 ,
 μ , μ μ
 μ , μ μ
 . μ μ
 μ (μ , 2001).
 , μ μ μ
 « » , 18° ,
 μ μ μ (*Lycopersicon esculentum* = μ
).
 1818.
 μ , μ
 , μ μ μ *var*

cerasiforme, μ μ () *L.pimpinellifolium*,
μ μ , μ .

2.2 - μ

μ
μ μ μ μ μ
. μ
19° ,
. μ . *Lycopersicon*
μ μ μ (2 =24)
. *Lycopersicon esculentum* ill ,
μ μ . Rick (1950),
| μ μ
| , μ μ . ,
Lycopersicon μ
μ μ μ μ μ . μ μ
μ μ μ μ , μ
μ μ μ μ μ
μ μ . μ μ μ
μ μ 50 , μ μ μ μ
(), μ μ μ μ
.
, μ
μ « μ ».
μ μ μ μ μ
μ μ μ μ .
, μ μ μ
μ .

μ μ . μ (μ)
 , μ , 20
 , 1+1=3.
 μ μ μ μ
 μ , μ , μ
 μ . μ μ μ
 . ,
 μ μ (, 2007).

2.3 μ

μ :) μ μ
 μ ,) , μ , μ , μ , ,
 μ μ ,)
 μ . μ
 μ ,) μ ,)
 ,) μ ,
) μ μ μ
 μ (long life semi long life).
 μ μ ,
 μ μ μ μ (non
 ripening genes). μ μ
 μ , μ , ,
 μ μ , μ

. , ,
 , μ μ
 (μ , 2001).
 | μ μ
 μ μ .
 | μ , μ
 μ , , , μ
 μ (μ , 2001).
 (F1) , μ .
 | , μ () μ μ
 μ . ,
 μ ,
 μ
 | μ , μ μ μ
 μ (μ , 2001).
 μ μ μ μ
 | μ μ μ μ .
 μ μ μ μ .
 | μ μ μ μ
 μ , μ
 μ
 μ . , μ
 | μ , (,
) (, 2007).

2.4

2.4.1

, , .

2.4.2

μ , , μ μ
μ μ , μ , μ
μ μ μ , μ μ μ
μ " "

, μ μ , μ μ μ
μ μ , μ μ μ μ
μ μ , μ μ μ
μ . μ μ μ μ ,
μ μ μ μ μ

μ , . μ μ μ μ

μ μ . μ
μ μ , ..
μ μ , μ
μ μ , .. μ
(2) , μ , .. (μ ,
2001).

μ (μ), ,
 . , μ
 | μ , μ
 μ . μ
 μ μ
 μ . μ
 . μ μ

2.4.5 -

μ μ 2-3/ μ 20
 . μ μ μ
 6-8 . μ
 μμ μμ , μ .
 | μ , 5 ,
 μ 5 μ 5 μ , μ
 μ μ μ μ μ ,
 μ (8). (2-7
) .

2.4.6

μ , μ μ .
 | μ μ () , μ 3, 4,
 5 μ μ (. 9)
 (μ , 2001).

2.4.7.

, μ , μ -
 μ , μ
 (μ) (10). μ
 μ , μ 3-5 . ()
 μ , μ μ .
 μ
 4 μ μ , μ
 μ μ μ μ
 , 10 . μμ
 " " 450 μ (μ , 2001).

2.5 μ μ

μ μ . μ
 μ μ
 | μ .

2.6 μ

μ μ μ
 μ
 ,
 μ . μ μ
 μ μ

μ
 (, 1997).

2.6.1

μ μ
 μ μ μ
 (1985) μ μ
 μ μ μ μ
 200 ppm N 350 ppm, μ
 μ μ μ μ
 (, 1995).

μ μ
 (, 1994). μ
 μ μ μ μ μ μ μ μ (μ ,
 2001).

μ μ
 μ μ μ μ , μ μ
 μ μ μ μ μ μ
 (, 1995).

μ μ μ μ μ μ μ μ μ μ
 μ μ μ μ μ μ μ μ μ μ μ μ
 (, 1994).

2.6.2

μ μ μ
μ μ
p
μ (Groot *et al.*, 2002).
μ μ
μ μ (Passam *et al.*, 2007).
μ , μ
μ μ
(, 1995). μ μ
μ μ μ μ Chapagain Wiesman (2004)
, Mg
Fe , μ
(Groot
et al., 2001)
μ (Groot *et al.*, 2002).
μ μ μ
2001). μ μ μ
μ Co (Chatterjee and Chatterjee, 2002) Zn
(Kaya and Higgs, 2002). , μ
μ μ
μ μ (*et al.*, 1994). μ
μ μ μ μ μ μ
μ μ μ μ μ μ ,
μ .

2.6.3

μ μ μ
μ μ μ μ

(Chapagain and Wiesman, 2004).
 (Chen and Gabelman, 2000; Rubio *et al.*, 2006)
 (Wang *et al.*, 2002).
 NaCl
 (Walker *et al.*, 2000).
 (100-120 ppm) (Lingle *et al.*, 1969).
 (200-300 ppm)
 (Lacatus *et al.*, 1994),
 Ca, K, Mg,
 (Ho *et al.*, 1993, Saure 2001, Navarro *et al.*, 2005).

2.6.4

(del Amor and Marcelis, 2006).
 (Blossom end rot, R) (11)
 (Ho *et al.*, 1993; Grattan and Grieve., 1999).
 Ca, K, Mg,
 (Ho *et al.*, 1993, Saure 2001, Navarro *et al.*, 2005).

NFT

μ μ

6 mM

μ μ

μ μ (Hao and Papadopoulos, 2003,2004)

2.6.6

μ

(, 2012).

2.6.7

μ

μ

pH

μ

pH

6-7 (Islam *et al.*, 1980; Akl *et al.*, 2003)

μμ

μ

μ

μ

μ

(Sonneveld, 2002).

μ

μ

μ

μ

μ

μ

μ

(Fernandez and Ebert, 2005; He *et al.*, 2005).

μ

μ

μ

μ

μ

μ

μ

(Fernandez and Ebert, 2005; He *et al.*, 2005).



13: μ .

2.6.8

μ 20 250 mg kg (Mills & Joens, 1996).
 μ μ μ μ μ
 Fe (Kaya and Higgs, 2002)

2.6.9

μ μ pH
 (Akl *et al.*, 2003) μ μ
 (Chaignon *et al.*, 2002) 5mg/l
 μ μ NFT.

(Bergman, 1988).

2.6.10

(Shenker *et al.*, 2004) Mils Jones
(1996) 250 mg kg

2.8

2002-2004, 13%, 8%, (7,5%), (7,5%) (5%).
20

20 μ μ ,
 . μ (6 μμ
 1982-84 28 μμ 2002-04), μ
 . 15, μ μ -
 1990 ,
 μ ,
 μ μ
 μ (μ μ). μ
 μμ , μ
 μ μ
 μ ,
 μ μ
 μ μ
 μ μ
 μ μ μ μ ,
 - μ ,
 μ μ
 60-80 / 10-20 /
 . , μ
 μ μ ,
 μ μ ,
 μ μ μ μ
 μ (μ , μ .)
 μ . μ μ ,
 .
 μ μ μ μ 1995-2000,
 μ
 . ,
 1978/79, μ 2005/06 83%, 486%,
 239% 361% (World Processing Tomato
 Council) (, 2007).

.
 ,
 .
 .
 8-10%
 .
 &
 2005 -
 34.849
 292.088 (, 2007).
 28.392 6.457
 .
 180.000 580.000 (μ
 3,2). 32.000
 320.000 (μ μ μ 10
). μ μ μ μ
 , , , , .
 μ μ μ μ μ , μ
 μ μ μ μ μ
 μ .
 130.000 (2005) μ μ μ 6,5
 . μ 25% (,) 15%
), 60% (,) 15%
 . μ μ μ μ
 (μ 60%)
 (40%). μ μ μ (μ
 70%) μ μ (30%). μ μ 25 -
 μ μ , 4 60%
 (, 2007).

2.9

μ

μ μ μ μ

, μ

μ μ

(de Kreij, 1995). μ μ

, μ

μ μ μ μ

μ - μ - , μ μ μ μ

μ , μ μ μ μ

μ (Resh,

1997). μ

μ

μ (, 2008).

μ μ μ μ

μ , μ μ μ

μ μ (μ μ

). μ , μ μ

μ (, 2007). , μ μ

, μ μ

μ μ

μ μ (μ , 1995). μ μ

μ μ μ μ

0-8 mm

μ (Gizas and Savvas, 2007).

μ μ ;

i. μ μ (. .

μ) μ ,

- ii. μ μ μ
- iii. μ , μ μ μ μ (. .) ,
- iv. μ μ μ μ μ ,
- v. μ (. .) μ μ μ ,
- vi. μ μ μ μ μ μ ,

μ , μ , μ , μ .
 μ , μ , μ , μ , μ .
 μ , μ , μ , μ , μ .
 μ , μ , μ , μ , μ .
 μ , μ , μ , μ , μ .
 μ , μ , μ , μ , μ .
 μ , μ , μ , μ , μ .
 μ , μ , μ , μ , μ .
 μ , μ , μ , μ , μ .

0-8 mm 2-10 mm.

μ μ μ μ μ (),
 μ μ μ .



14: μ μ μ μ .

μ μ μ ,
 μ μ μ . μ μ
 μ . μ .
 μ μ μ μ μ
 μ . 60 μ , μ ,
 μ 0-8, μ 2-10 .
 μ μ μ μ μ
 μ μ μ μ μ
 μ μ μ μ μ
 (10x10 cm), μ μ ,
 .



15:

μ

μ

3.1.2.

μ

μ

μ

μ

μ

μ

μ

μ

μ

,

μ

μ

14/3/2012 μ

μ

μ

μ

μ

μ

μ

μ

1/100 μ

μ

μ

μ

μ

μ

μ

,

μ

μ

μ

(

μ

μ

Crison Cm25)

pH

(μ

Crison pH 25) μ

(HNO₃)

	μ					
E_c	2,8	dS/m	2,6	dS/m	2,65	dS/m
pH opt.	5,6	mmol/l	5,6	mmol/l	5,6	mmol/l
[K]	8,4	mmol/l	7	mmol/l	8,5	mmol/l
[Ca]	11,5	mmol/l	5,1	mmol/l	4,5	mmol/l
[Mg]	5,5	mmol/l	2,4	mmol/l	2,5	mmol/l
[NO₃⁻]	17	mmol/l	14,3	mmol/l	14,25	mmol/l
[NH₄]	1	mmol/l	1,5	mmol/l	1,5	mmol/l
[H₂PO₄⁻]	1,25	mmol/l	1,5	mmol/l	1,5	mmol/l
[Fe]	20	μ mol/l	15	μ mol/l	15	μ mol/l
[Mn]	10	μ mol/l	10	μ mol/l	10	μ mol/l
[Zn]	5	μ mol/l	5	μ mol/l	6	μ mol/l
[Cu]	0,7	μ mol/l	0,8	μ mol/l	0,8	μ mol/l
[B]	40	μ mol/l	35	μ mol/l	35	μ mol/l
[Mo]	0,5	μ mol/l	0,5	μ mol/l	0,5	μ mol/l

3:

μ

,

.

μ μ)

(μ Schneider CCT15720 IHP),

μ

μ μ ,

μ

μ

(33ml μ 1min).)

μ

μ μ μ .)

μ μ

μ 25

20

μ

μ

μ

μ

, 16.

μ μ

μ

' μ ,

8 m³/sec.

μ

μ

μ

μ μ

μ

.

μ

μ (5-15). μ
 μ (, 2007).
 μ (200 lit),
 μ , μ
 μ 190 . μ
 μ μ μ μ
 μ μ 14 μ μ
 5 min.

3.1.3 μ

μ μ Primadonna
 (*Solanum lycopersicum* L., cv. Primadonna F1),
 . μ μ (selfgrafted) μ
 Maxi-Fort, Solanum.
 μ μ μ ,
 Fusarium, (TMV).
 15/3/2012 μ 180 μ , 6-7
 . μ μ μ
 μ μ μ . μ
 μ μ 20/7/2012.



16:

Primadonna.



17:

μ

μ



18: μ μ μ 2-

10 mm.



19: μ μ μ 0-

8mm.



20: μ μ .



21: μ μ μ .

μ μ μ , 20 μ μ μ
 μ .
 , μ μ .
 μ μ μ μ 12 .
 , μ ,
 , μ μ .
 μ μ μ μ
 μ μ . μ μ 65 C
 μ .
 μ , μ , μ
 .
 μ μ μ , .
 16/5/2012 μ μ 19/7/2012
 μ . μ μ
 , μ
 μ μ μ .
 μ μ μ .



22: μ
 μ
 .



23 & 24:
μ
Primadonna

μ μ μ

3.1.4

μ

μ

μ

μ

μ

-

trialeurodes vaporariorum.

μ 500 *macrolophus caliginosus* (KOPPERT)

μ

μ

μ

μ

μ

Oberon (

: spirimesifen)

μ

.

μ

μ

tuta absoluta.

μ

μ

μ

μ

.

μ

μ

μ

4 gr

μ

μ Trigard (

: cyromazine)

μ μ

(Patakioutas *et al.*, 2007).



25:

μ

μ

μ

.

3.2

3.2.1

μ

μ 2/7/2012 μ

(μ , ,)

μ μ μ .
μ 5-8 μ μ ,

— μ

μ μ μ μ μ Minolta μ CR
200, μ μ μ μ μ L*, a*,
b* μ , y, z μ
CIE. μ L* μ μ μ 0 (μ) 100
(), a* μ μ
μ , b* μ μ μ μ
μ .
μ μ μ μ μ
μ μ μ μ .
μ μ μ μ μ 5-8 μ ,
μ .

—————

μ μ μ μ
Chatillon DFIS 10 μ μ Chatillon Tcm 201,
μ μ μ μ 6,3 mm, μ
200 mm/min, μ μ 50 N μ 0,1 N.
μ μ μ μ
μ μ

3.2.2 μ μ

20/7/2012, μ μ μ μ
μ μ . μ
μ , μ
: K, Ca, Mg.

_____ μ μ
μ , μ μ
μ . μ ,
μ
μ ,
μ μ μ . μ μ
μ μ μ 0.5mm.

| _____
0.5 gr μ μ
μ μ .
8 500 C. μ 24
μ μ
μ 10ml HCl 1 . μ ,
μ μ μ
μ 100ml, μ
μ . μ μ μ
100ml , μ μ
μ μ 40 .
1:100 μ , μ 100 ml
μ , μ
μ .



17: μ μ μ

μ

μ , μ μ μ
 μ (Atomic Emission Spectroscopy AES). μ μ
 μ μ μ μ μ μ
 μ μ .

μ μ μ

μ μ μ Ca Mg, μ
 μ μ (Atomic Absorption Spectroscopy
 AAS). μ μ μ
 μ μ μ .

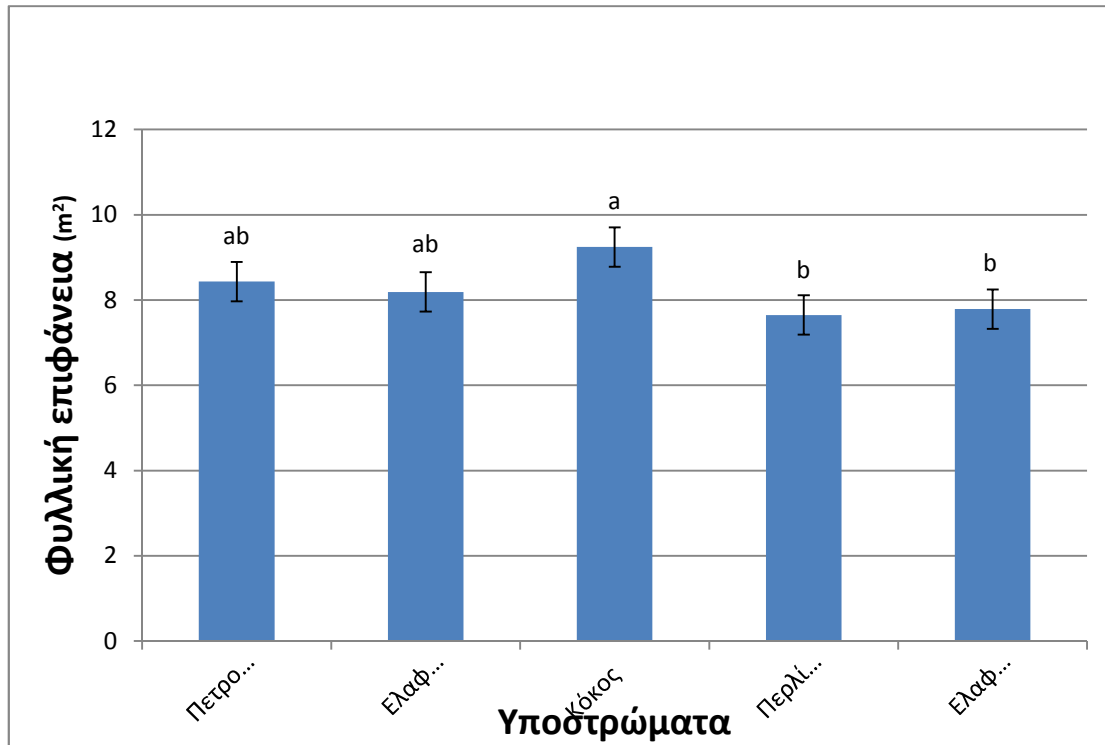
μ
μ
μ
μ
/ μ . μ
μ μ
μ μ
μ μ

μ , μ
 μμ μ μμ
 STATISTICA. μ μ
 μ μ μ
 μ μ . μ
 μ μ μ Duncan,
 μ 5%.

4.1 μ μ

μ , μ ,
 μ μ μ μ .
 μ μ μ μ .
 μ μ μ
 μ μ μ .
 μ , μ μ μ .
 μ , μ μ
 (%) .

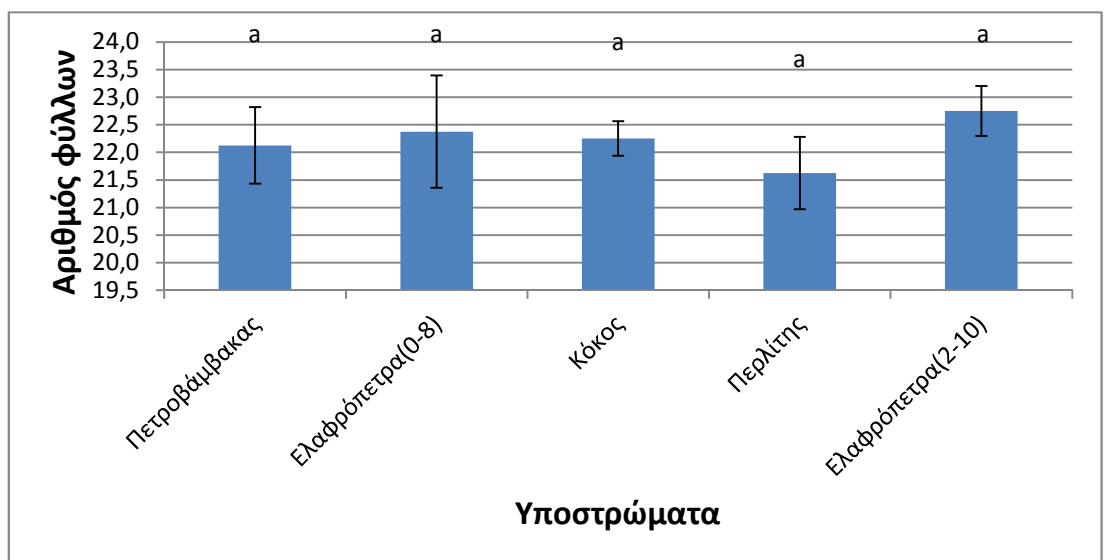
μ μ μ (2-10),
 μ μ μ .



μ 2:
 μ , μ μ , (0-8),
 (2-10), .

4.1.3 μ

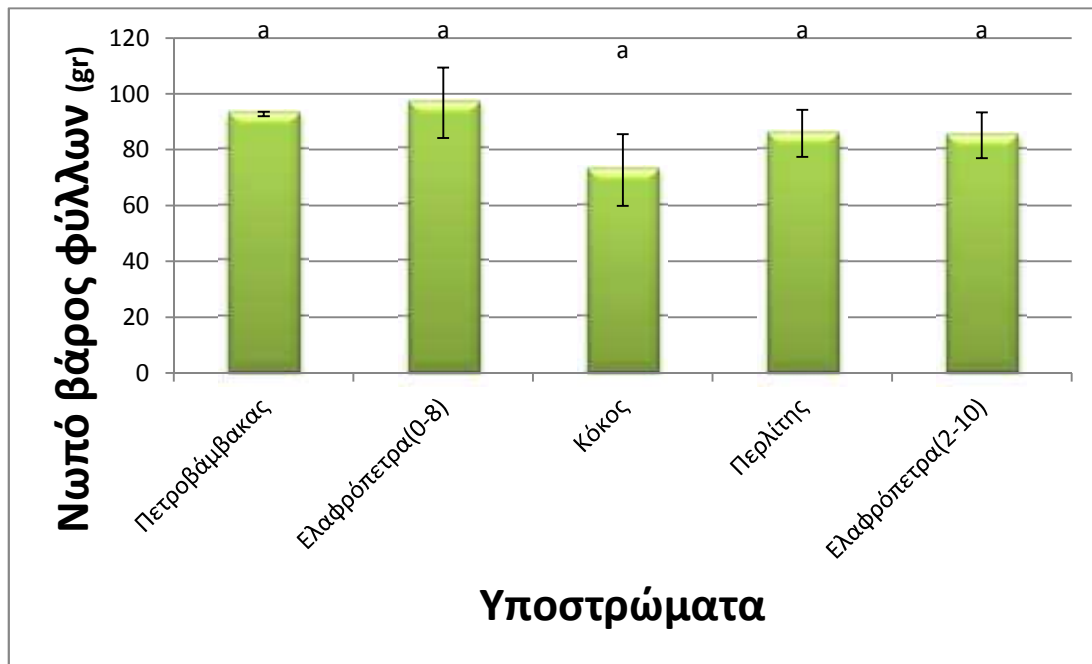
μ μ μ ,
 μ (μ 3)
 μ μ μ



μ 3: μ μ μ (0-8), (2-10), .

4.1.4

μ 4 μ 17/5/2012. μ μ μ , μ μ . μ μ μ , μ μ μ μ μ μ μ μ μ μ .



μ 4:

μ

μ

(0-8),

(2-10),

μ

μ

μ

μ

μ

(65 C),

μ

μ

μ 5,

μ

μ

μ

(0-8)

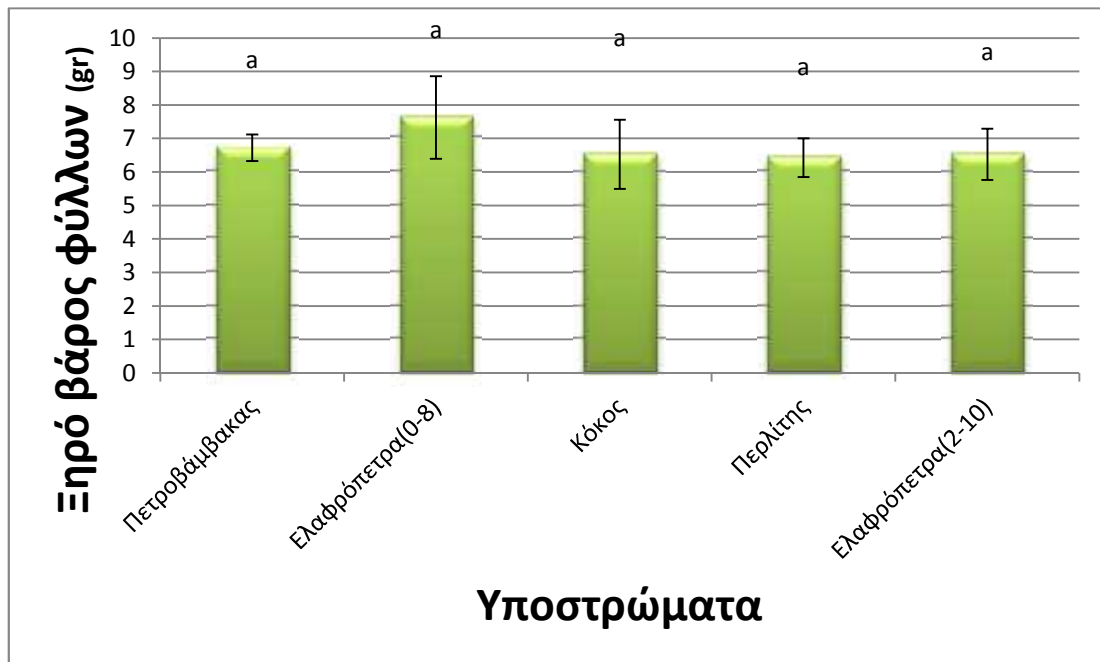
μ μ

μ

μ μ

μ

μ .

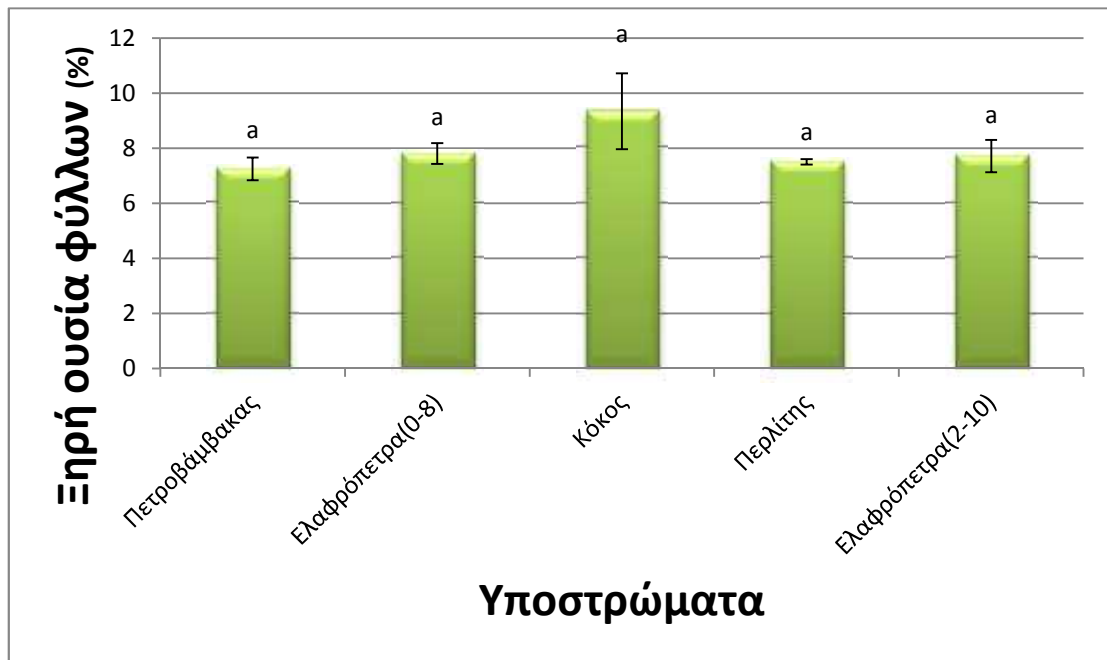


μ 5:

(2-10), μ , μ (0-8),

4.1.5

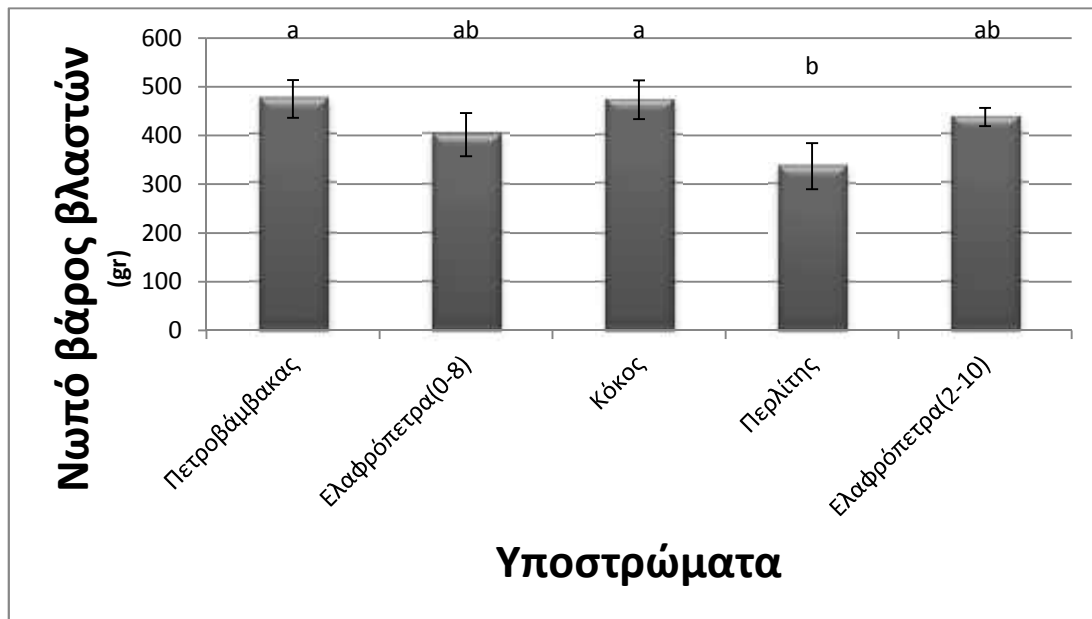
μ 6
 μ , μ
 μ μ . μ
 μ μ Duncan.



μ 6:
 (2-10), μ , (0-8),

4.1.6

μ . μ μ 7,
 μ μ , μ μ ,
 μ μ , μ μ ,
 μ , μ μ , μ



μ 7:

μ

μ

μ

(0-8),

(2-10),

μ

μ 8

μ

μ

90%

μ

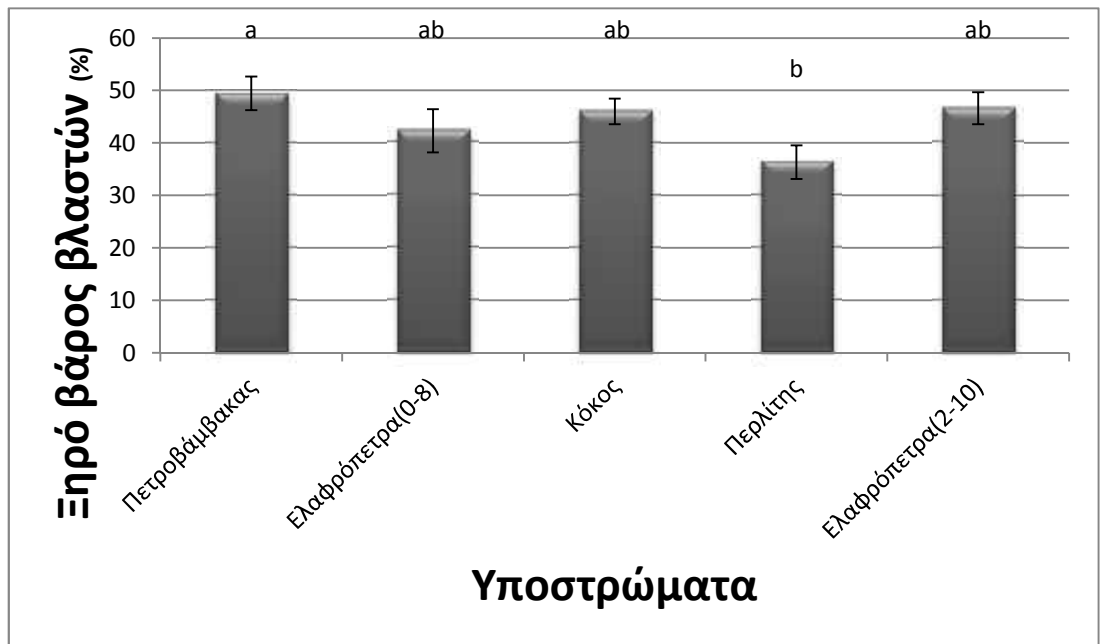
μ

μ

μ

μ

μ .



μ 8:

μ

μ

μ

(2-10),

(0-8),

4.1.7

μ .

μ

μ 9,

. μ

μ μ 9,8%

11%

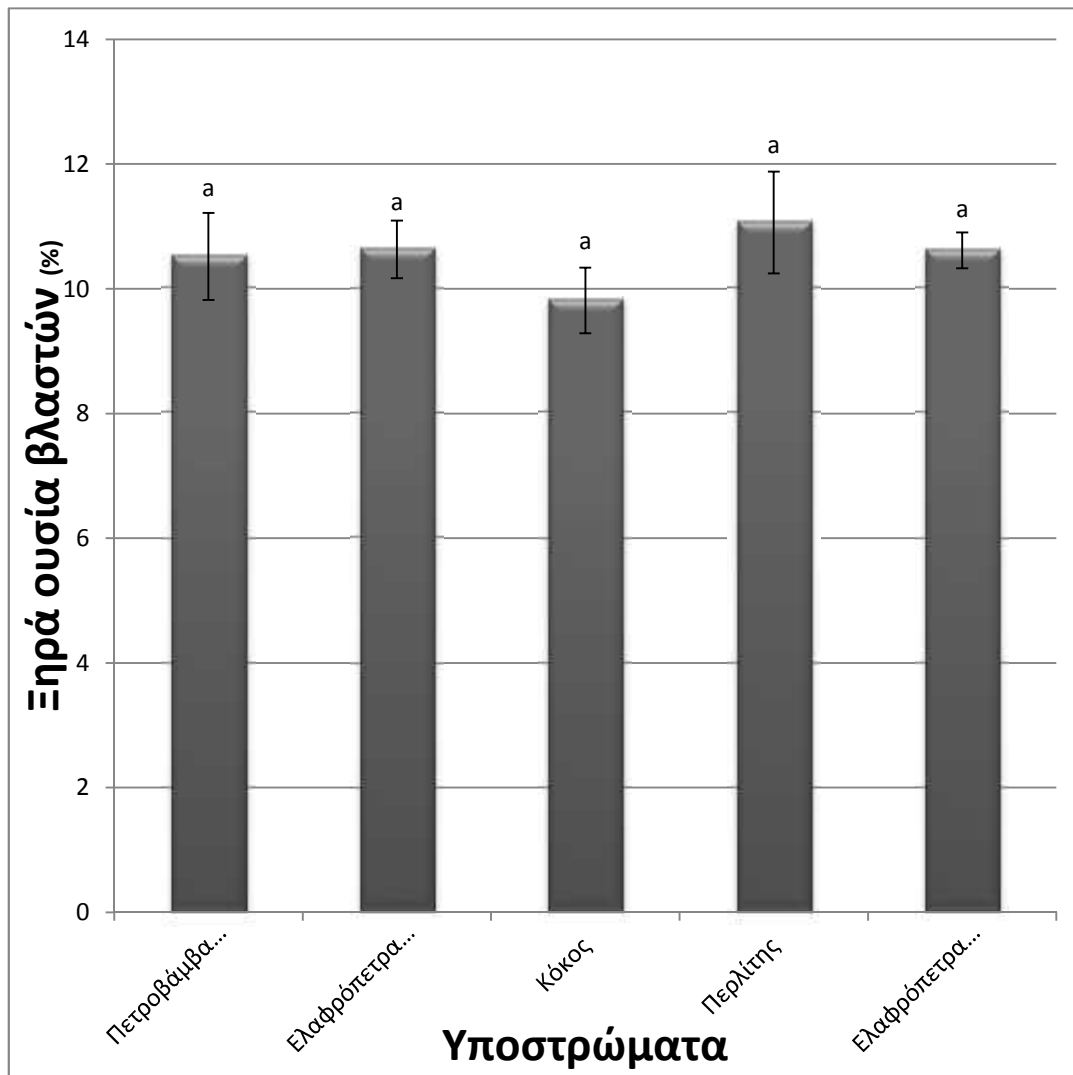
μ μ μ μ

μ μ .

μ ,

μ

μ .



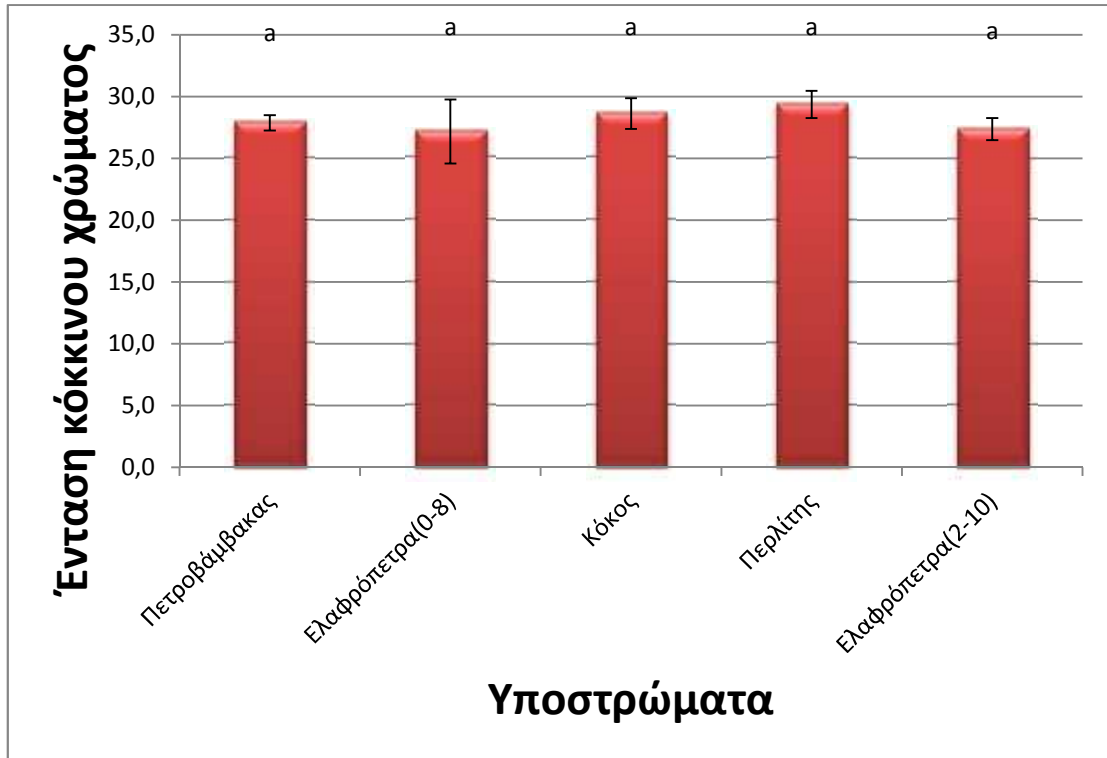
μ 9: μ (%) μ μ
 (0-8), (2-10),

4.2

μ

μ μ μ (a*) μ μ μ μ μ μ μ μ

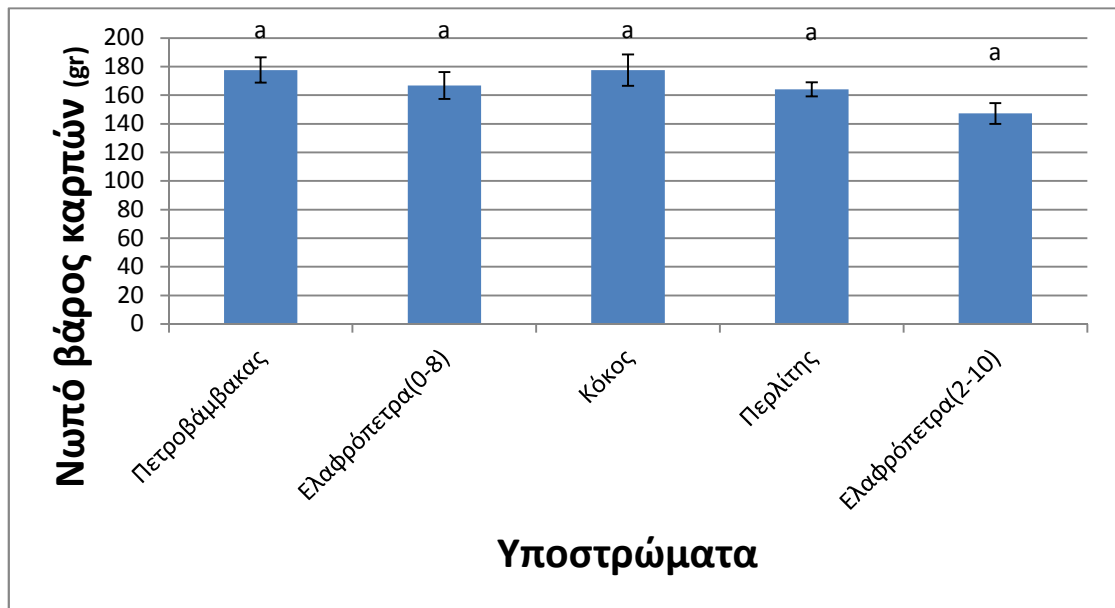
μ (μ 10). Duncan
 μ μ
 μ , μ μ
 .



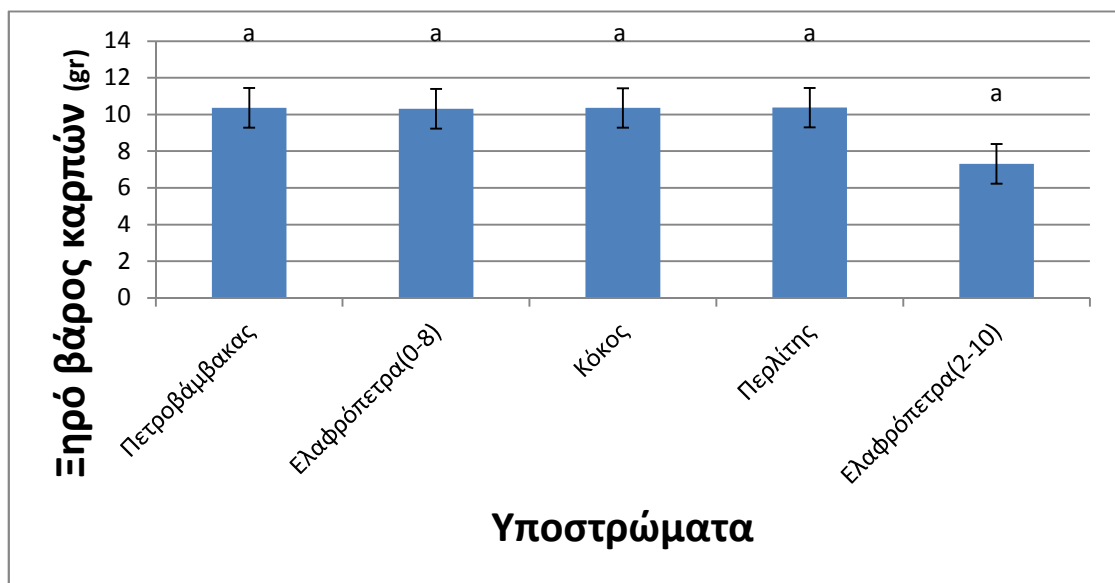
μ 10: μμ
 μ (a*) μ μ μ
 (0-8), (2-10), .

4.2.1 - μ

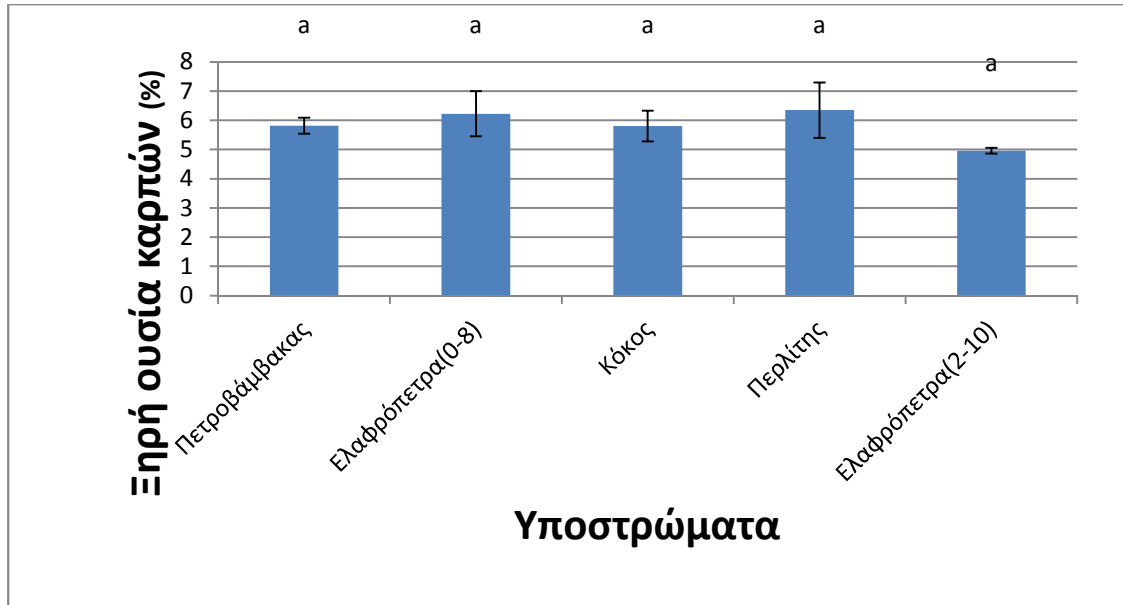
μ μ
 , μ
 μ . μ 11 12
 μ . (%)
 μ 13. μ ,
 μ .



μ 11: μ μ (0-8), μ (2-10),



μ 12: μ μ (0-8), μ (2-10),



μ 13:

μ

μ

(0-8),

μ

(2-

10),

4.2.2

μ 14

(Brix)

μ

.

μ

μ

μ

,

,

μ

,

μ

,

μ (Hardy, 2010),

μ

μ

μ

μ

.

μ

μ

μ

μ

,

μ

Brix

μ

μ

(Seacheol *et al.*, 2002).

μ

,

μ

μ

μ

.

μ

μ

μ

(Davies and Hobson, 1981).

μ

μ

μ

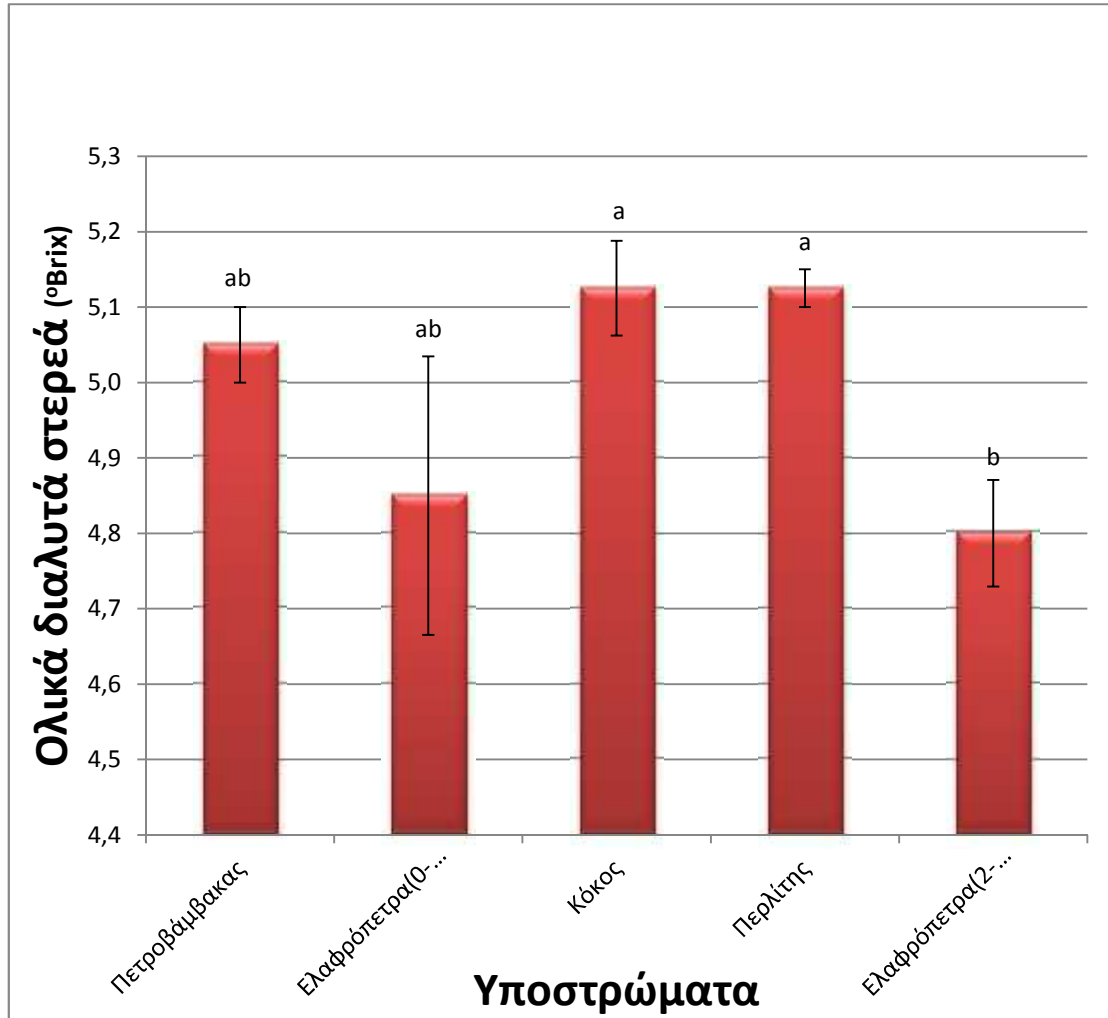
(0-8) μ μ

5

4,8

,

μ 2-10 μ 4,6. μ

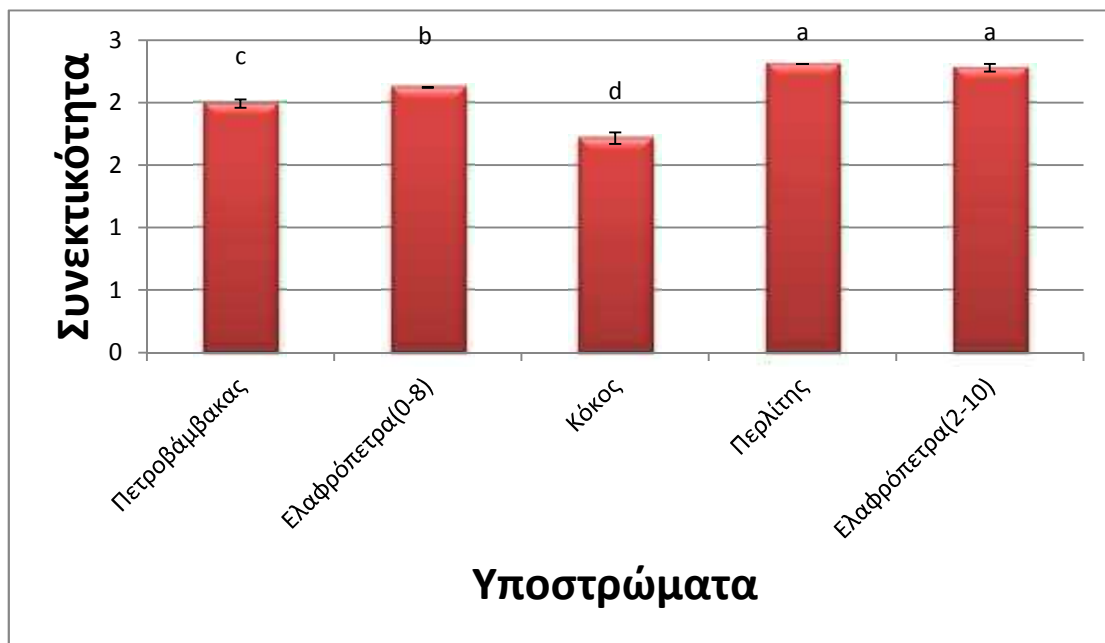


μ 14: μ μ μ (0-8), (2-10),

4.2.3

μ 15 () μ μ μ μ μ μ

(Artes *et al.*, 1999). μ μ μ
 Duncan, μ μ
 (Najib *et al.*, 1997). μ μ 2,3
 (2-10), μ
 μ , μ μ μ
 μ μ μ (0-8)
 μ , μ μ μ
 μ μ μ μ
 μ .



μ 15: μμ
 μ , μ
 (0-8), (2-10),

4.3

μ

μ

4.3.1

μ

μ

μ

μ (μ 16 17)

μ

μ

μ

3000 gr

μ

23-24

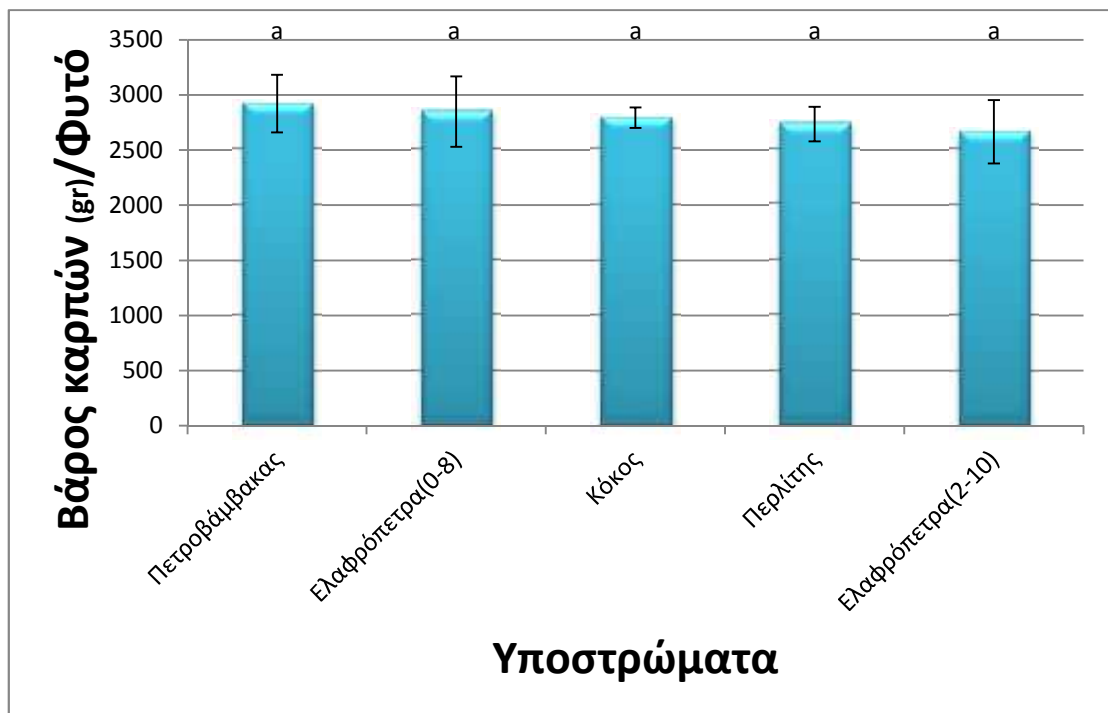
μ

μ

μ

μ

μ



μ 16:

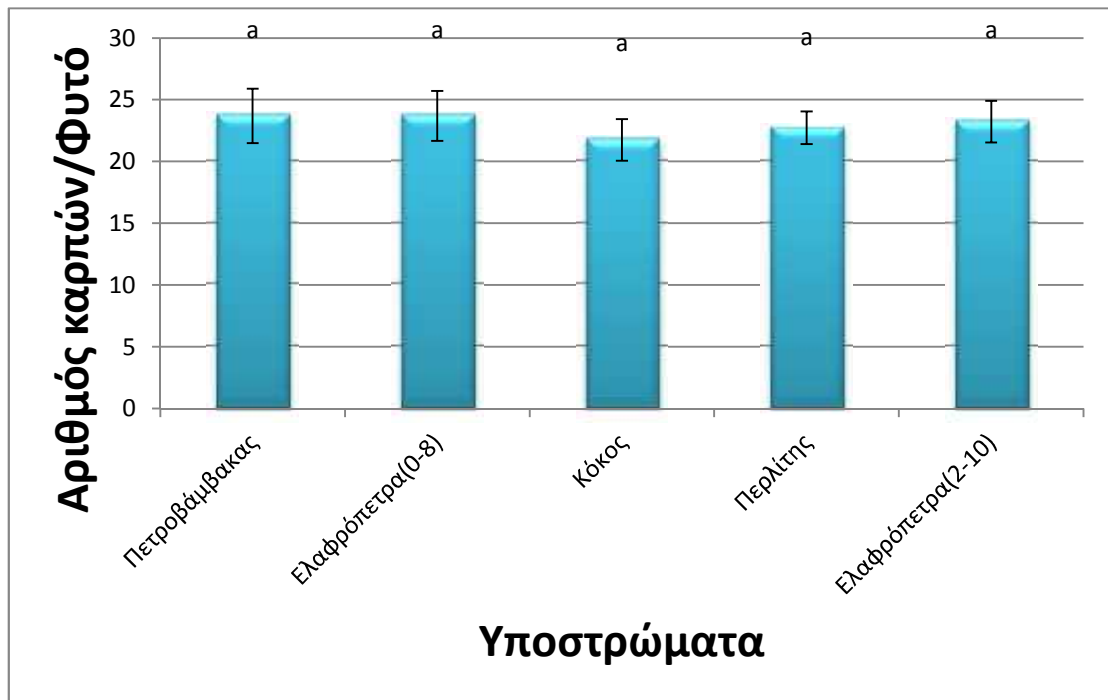
(gr)

μ

μ

(0-8),

(2-10),



μ 17:

μ

μ

μ

(0-8),

(2-10),

.

4.3.2

μ

μ .

μ 18,

μ

μ

μ .

μ

μ

μ

μ .

μ ,

(0-8)

, μ

,

(2-10)

μ

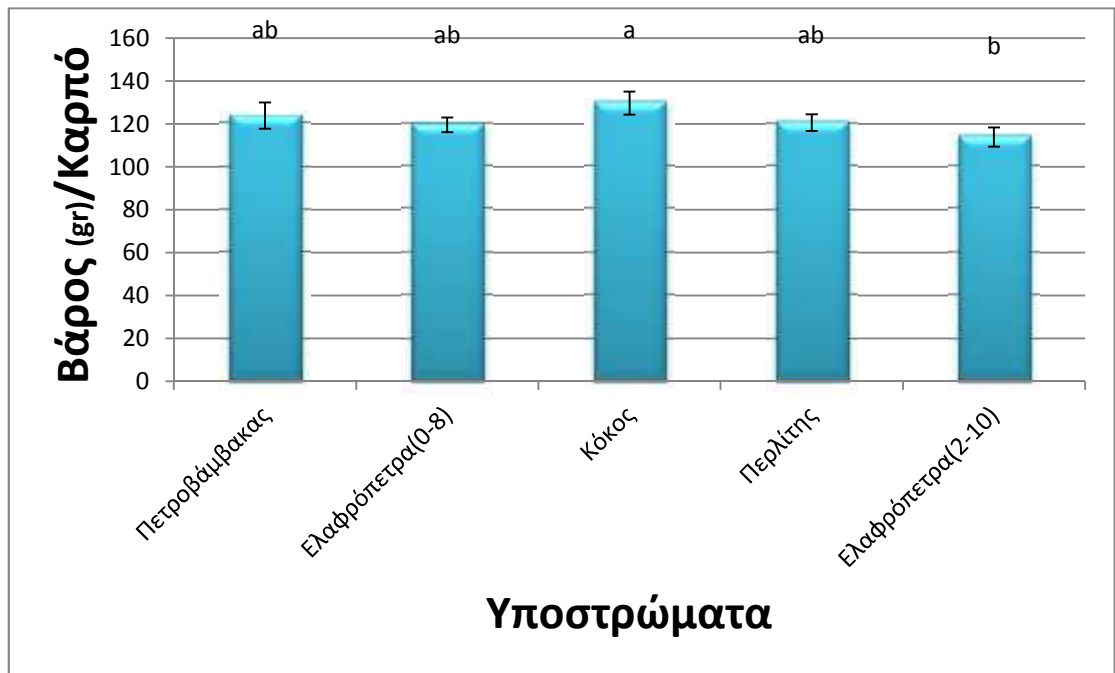
μ

μ

,

μ

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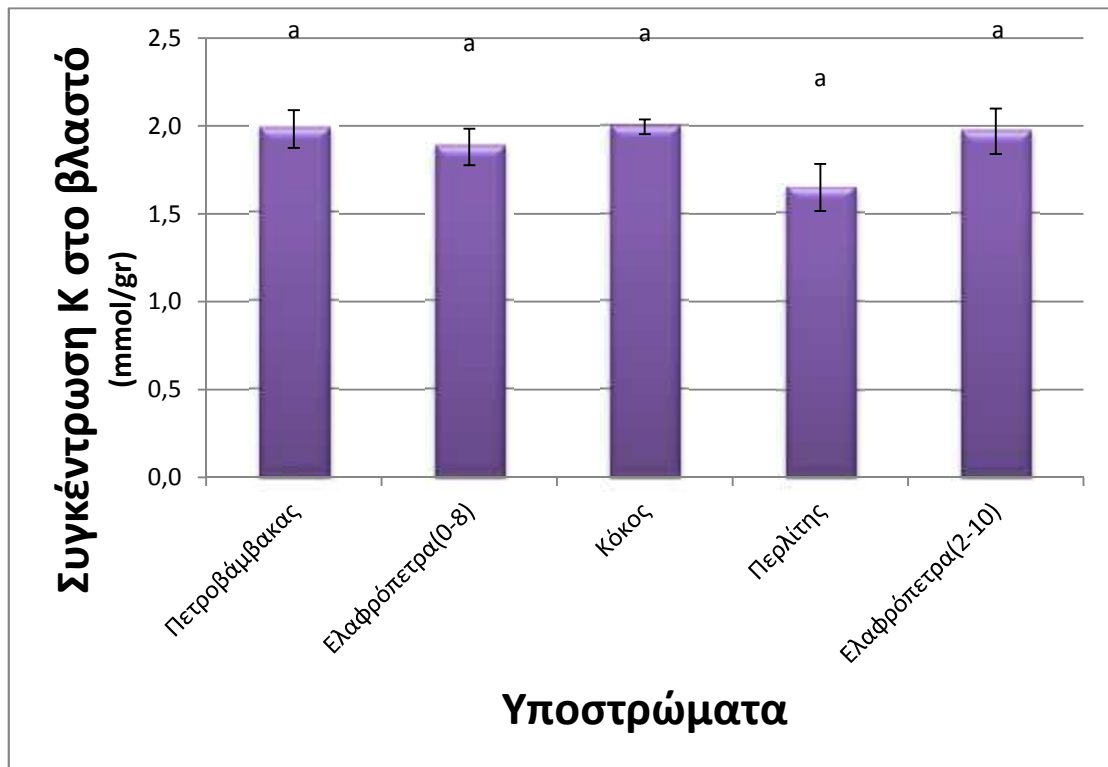


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4.4.1

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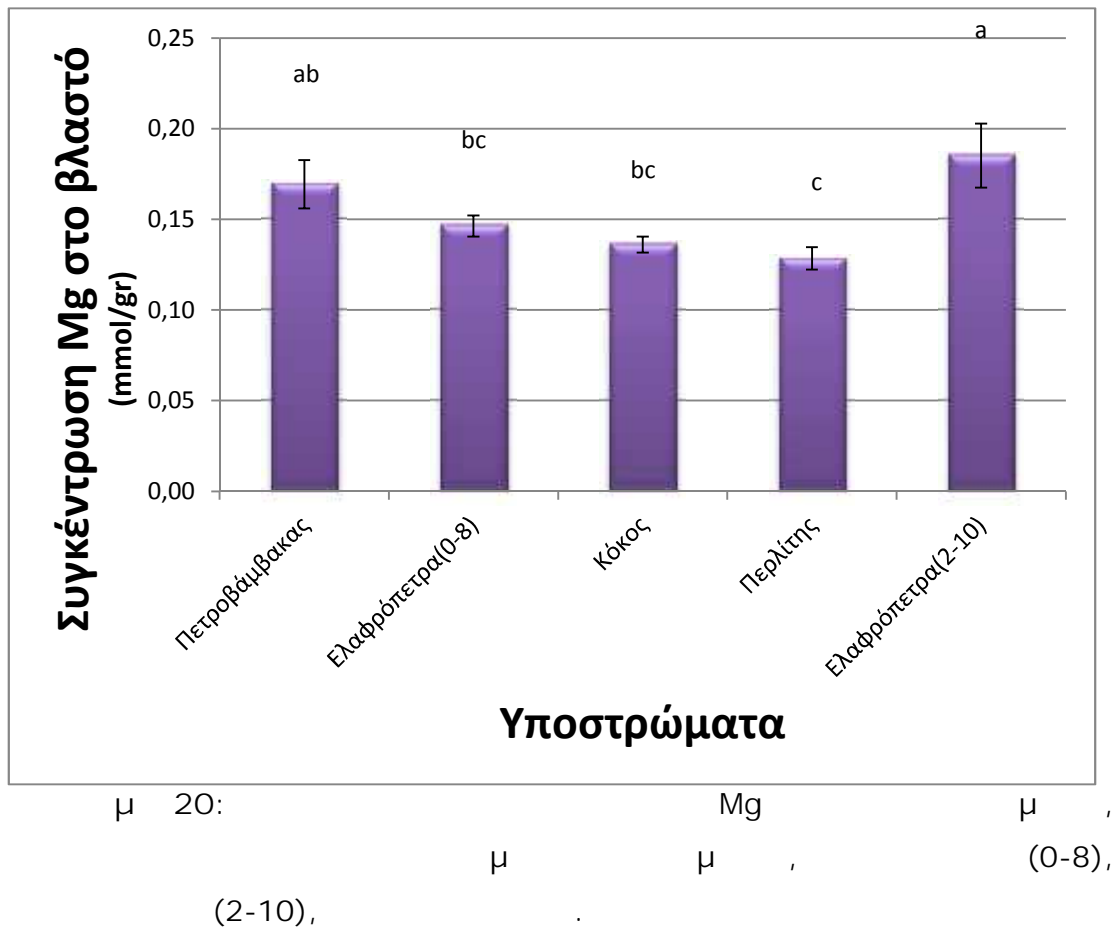
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4.4.2

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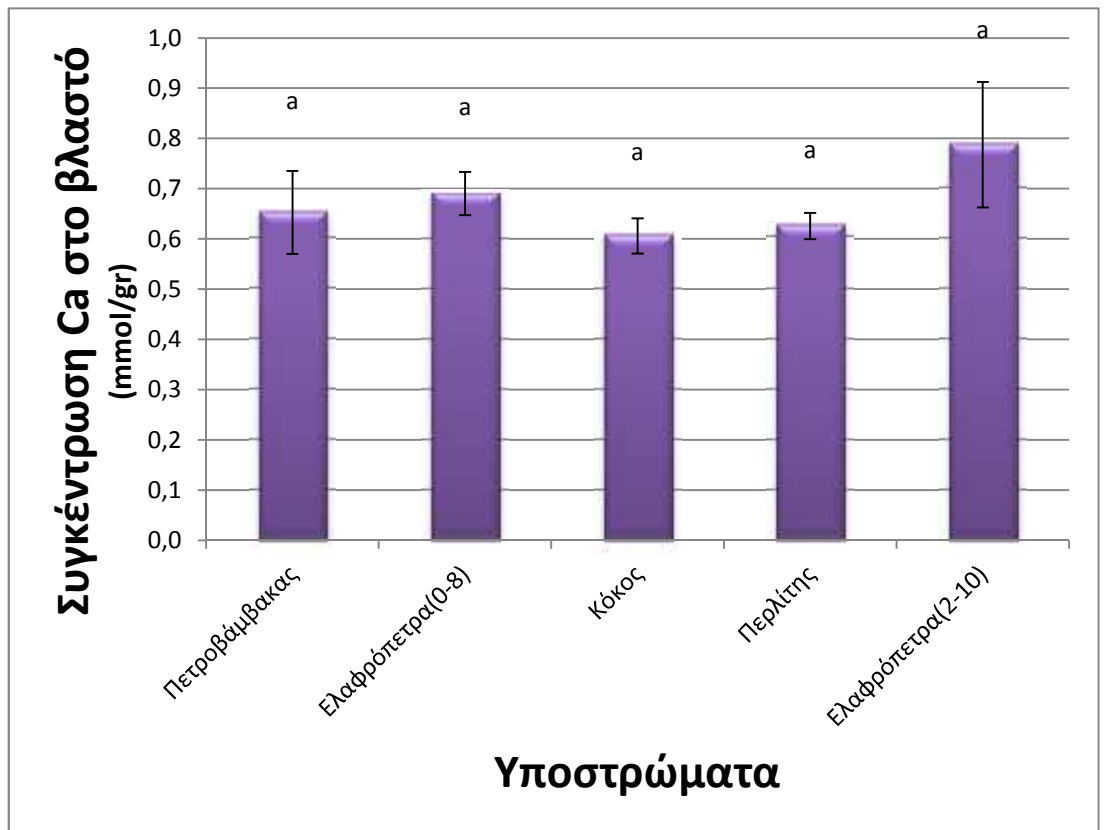
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4.4.3

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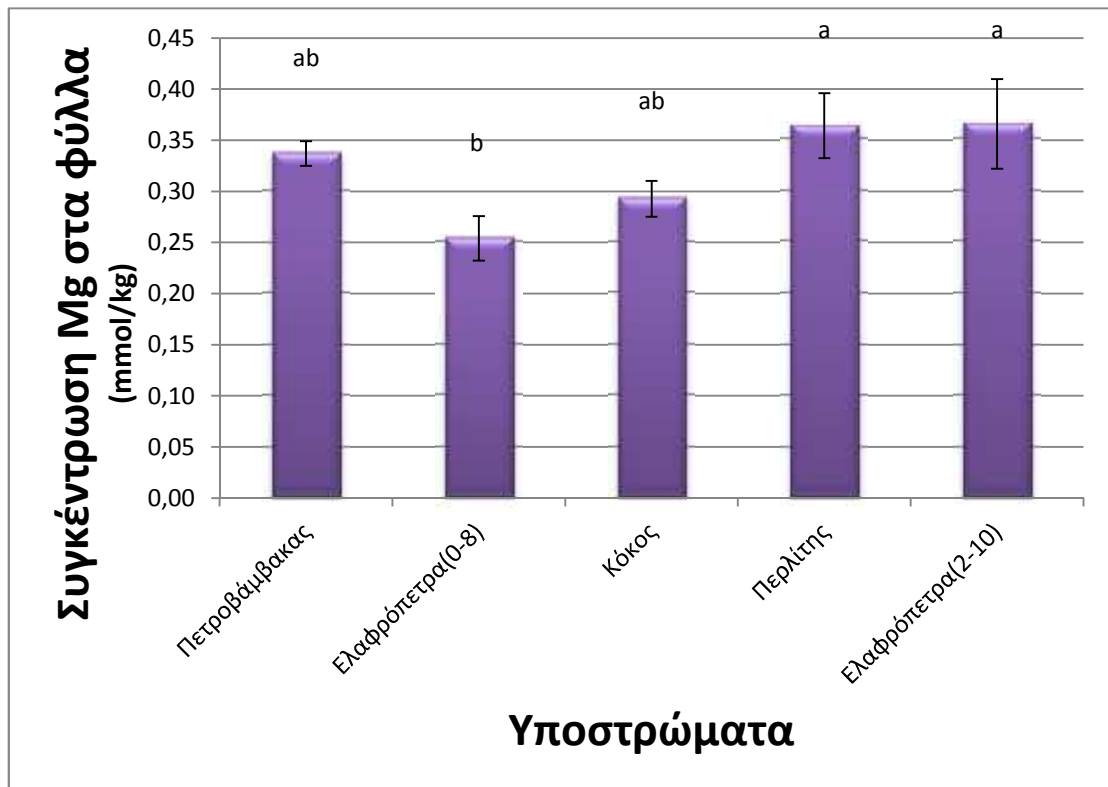
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4.4.4

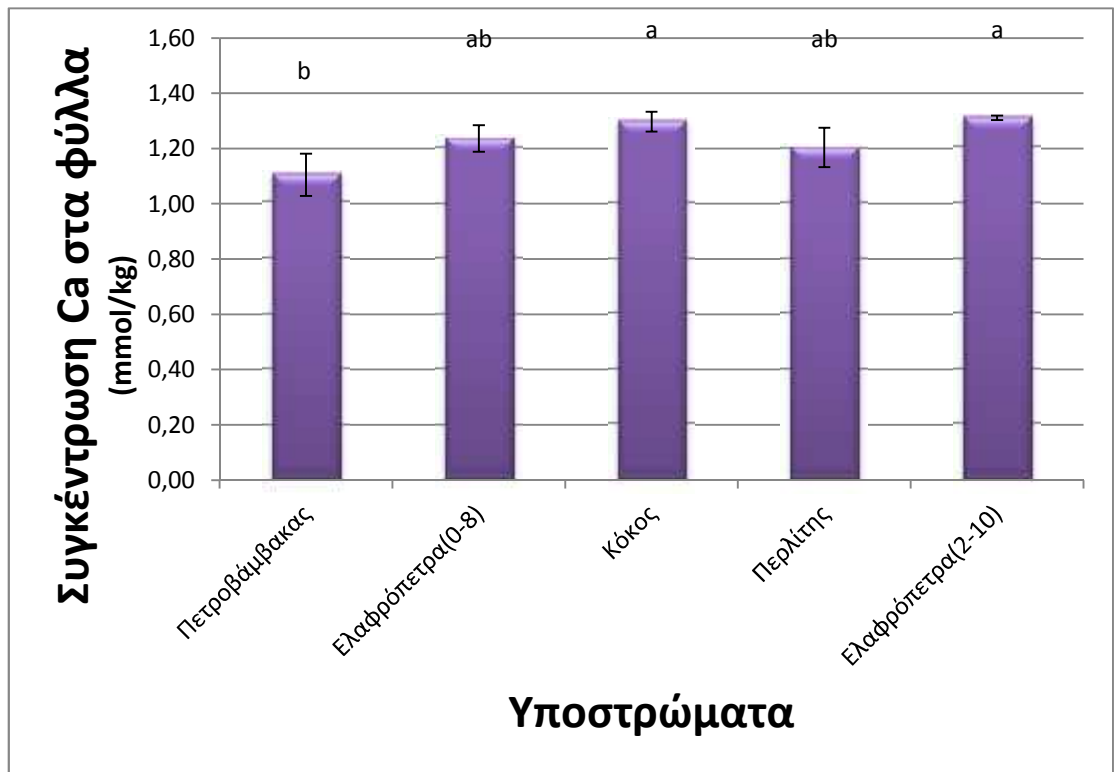
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μ 23: Mg μ , (0-8), (2-10),

4.4.6 Ca

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